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Key indicators

Single-crystal X-ray study T = 273 K Mean σ (C–C) = 0.004 Å R factor = 0.045 wR factor = 0.120 Data-to-parameter ratio = 7.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

(*R*)-2,2'-Bis(methoxymethoxy)-1,1'-binaphthalenyl-3-carbaldehyde

The title compound, $C_{25}H_{22}O_5$, was obtained from (*R*)-2,2'bis(methoxymethoxy)-1,1'-binaphthalenyl by *ortho*-lithiation and quenching with *N*,*N*-dimethylformaldehyde. The molecular packing in the crystal structure is stabilized by weak $C-H\cdots O$ hydrogen-bonding interactions. Received 17 February 2006 Accepted 2 March 2006

Comment

The title compound, (I), is an intermediate of chiral salicylaldehyde derivatives, for subsequent conversion to chiral Schiff base used in enantioselective sulfoxidation (Zeng, Wang, Wang *et al.*, 2005; Zeng, Wang, Weng *et al.*, 2005). Bond lengths and angles are in agreement with values reported for a similar compound (Tachi *et al.*, 1999). The dihedral angle between the planes of the naphthalene ring systems is 74.92 (6) Å.





Figure 1

The molecular structure of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 30% probability level. H atoms are drawn as spheres of arbitrary radii.

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organic papers

The molecular packing in the crystal structure is stabilized by weak $C-H \cdots O$ hydrogen-bond interactions (Table 1 and Fig.1).

Experimental

To a solution of (R)-2,2'-bis(methoxymethoxy)-1,1'-binaphthalenyl (4.30 g, 11 mmol) in dry tetrahydrofuran (40 ml) in an ice–salt bath, an LiBu (7 ml, 1.6 *M*) solution in hexane was added. To the resulting slurry in the ice–salt bath, *N*,*N*-dimethylformaldehyde (0.9 ml, 12 mmol) was added after 3 h. After a further 3 h, a saturated aqueous NH₄Cl solution was added to quench the reaction. The organic layer was extracted with ethyl acetate, dried over anhydrous MgSO₄ and concentrated under vacuum. The residual liquid was purified by column chromatography to give 1.74 g of the title compound in 39% yield. Single crystals were obtained from a petroleum ether and ethyl acetate (10:1) solution by slow evaporation.

Mo $K\alpha$ radiation

reflections

 $\theta = 2.2-27.3^{\circ}$ $\mu = 0.09 \text{ mm}^{-1}$

T = 273 (2) K

 $R_{\rm int} = 0.023$

 $\theta_{\rm max} = 25.0^{\circ}$

 $h = -9 \rightarrow 10$

 $\begin{array}{l} k=-14\rightarrow 13\\ l=-22\rightarrow 19 \end{array}$

Block, colorless

 $0.45 \times 0.45 \times 0.30 \text{ mm}$

2103 independent reflections

1991 reflections with $I > 2\sigma(I)$

Cell parameters from 5357

Crystal data

 $C_{25}H_{22}O_5$ $M_r = 402.43$ Orthorhombic, $P2_12_12_1$ a = 9.102 (2) Å b = 12.322 (3) Å c = 18.542 (4) Å $V = 2079.6 (8) \text{ Å}^3$ Z = 4 $D_x = 1.285 \text{ Mg m}^{-3}$

Data collection

Bruker APEX area-detector diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Bruker, 2001) $T_{min} = 0.959$, $T_{max} = 0.978$ 10490 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.061P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.045$	+ 0.3987P]
$wR(F^2) = 0.120$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.15	$(\Delta/\sigma)_{\rm max} = 0.002$
2103 reflections	$\Delta \rho_{\rm max} = 0.17 \ {\rm e} \ {\rm \AA}^{-3}$
273 parameters	$\Delta \rho_{\rm min} = -0.15 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

Table 1

			~	
H	/drogen-bon	d geometry	(Å.	°)
			·	

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$\begin{array}{c} \hline C7 - H7 \cdots O2^{i} \\ C12 - H12 C \cdots O1^{ii} \end{array}$	0.93	2.58	3.283 (4)	133
	0.96	2.55	3.354 (6)	142

Symmetry codes: (i) x - 1, y, z; (ii) $x + \frac{1}{2}, -y + \frac{1}{2}, -z$.



Figure 2

Packing view, showing the weak C-H···O hydrogen-bond interactions (dashed lines) linking the molecules. [Symmetry codes: (i) x - 1, y, z; (ii) $\frac{1}{2} + x$, $\frac{1}{2} - y$, -z.]

All H atoms were placed in geometrically idealized positions and constrained to ride on their parent C atoms, with C–H distances of 0.95 (aromatic), 0.97 (CH₂) and 0.98 Å (CH₃), with $U_{iso}(H) = 1.2_{eq}(aromatic and CH₂) or <math>1.5U_{eq}(CH_3)$. In the absence of significant anomalous dispersion effects, Friedel pairs were averaged.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996), *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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